organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.003 Å R factor = 0.035 wR factor = 0.083 Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-Chloro-1,3-dioxa- $2\sigma^3\lambda^3$ -phosphaanthracen-4-one

The molecule of the title compound, $C_{11}H_6ClO_3P$, is essentially planar, except for the P, Cl and carbonyl O atoms. Bond lengths at phosphorus are P-O = 1.6110(18) and 1.6290(18) Å, and P-Cl = 2.0890(9) Å. The molecules are linked to form ribbons parallel to the *b* axis by two $C-H\cdots O$ and one $P\cdots O$ interaction.

Comment

The title compound, (I), was obtained as a synthetic intermediate en route to phosphorus-substituted calix[4]arenes (Kunze, 2002; Kunze *et al.*, 2002).



The structure of (I) is shown in Fig. 1. Bond lengths and angles may be considered normal (Table 1). All non-H atoms, except P, Cl and O3, are coplanar (r.m.s. deviation 0.028 Å); these three atoms lie 0.491 (1), 2.525 (1) and -0.256 (2) Å, respectively, out of the mean plane (the minus sign indicating the opposite side of the plane).

The molecular packing involves three contacts, namely two 'weak' C-H···O hydrogen bonds (Table 2) and a P···O3 contact of 3.072 (2) Å (operator of O3: -x + 2, $y - \frac{1}{2}$, $-z + \frac{1}{2}$). The overall effect of these is to link the molecules in ribbons parallel to the *b* axis (Fig. 2). The Cl···Cl contact between ribbons is, at 3.897 (1) Å (operator $x + \frac{1}{2}, -y + \frac{1}{2}, -z$), probably too long to be regarded as significant.

The structure of an isomeric material is presented in the following paper (Jones *et al.*, 2002).

Experimental

The title compound was prepared by the reaction of 3-hydroxynaphthalene-2-carboxylic acid with phosphorus trichloride in toluene (Hoechst, 1966) and recrystallized from dichloromethane/diethyl ether (2:1 ν/ν) (Kunze, 2002).

Crystal data

C ₁₁ H ₆ ClO ₃ P	Mo $K\alpha$ radiation
$M_r = 252.58$	Cell parameters from 56
Orthorhombic, $P2_12_12_1$	reflections
a = 6.0031 (15) Å	$\theta = 10-11.5^{\circ}$
b = 10.0697 (15) Å	$\mu = 0.51 \text{ mm}^{-1}$
c = 17.028 (3) Å	T = 173 (2) K
V = 1029.3 (4) Å ³	Prism, colourless
Z = 4	$0.5 \times 0.3 \times 0.2 \text{ mm}$
$D_x = 1.630 \text{ Mg m}^{-3}$	

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Received 11 November 2002 Accepted 19 November 2002 Online 30 November 2002

Data collection

Stoe Stadi-4 diffractometer ω/θ scans Absorption correction: none 3936 measured reflections 2367 independent reflections 2114 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.038$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.083$ S = 1.062367 reflections 145 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

P-O1 P-O2	1.6110 (18) 1.6290 (18)	P-Cl	2.0890 (9)
O1-P-O2	100.95 (9)	C3-O1-P	123.43 (16)
O1-P-Cl	100.21 (7)	C11-O2-P	127.24 (15)
O2-P-Cl	98.33 (7)		

 $\theta_{\rm max} = 27.5^\circ$

 $h = -7 \rightarrow 3$ $k = -13 \rightarrow 0$

 $l = -22 \rightarrow 22$

3 standard reflections

+ 0.1959P]

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.22 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.26 \text{ e} \text{ \AA}^{-3}$

974 Friedel pairs

frequency: 90 min

intensity decay: none

 $w = 1/[\sigma^2(F_o^2) + (0.0289P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

Absolute structure: Flack (1983),

Flack parameter = -0.03 (10)

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} C1 - H1 \cdots O2^{i} \\ C8 - H8 \cdots O3^{i} \end{array}$	0.95	2.60	3.542 (3)	174
	0.95	2.56	3.484 (3)	164

Symmetry code: (i) $2 - x, \frac{1}{2} + y, \frac{1}{2} - z$.

The absolute structure was determined on the basis of 974 Friedel pairs. The bulk material is racemic, so that the concept of absolute configuration can only be applicable to the measured crystal. H atoms were included using a riding model with fixed C—H bond lengths of 0.95 Å; $U_{\rm iso}({\rm H})$ values were fixed at 1.2 times the $U_{\rm eq}$ value of the parent atom.

Data collection: *DIF*4 (Stoe & Cie, 1992); cell refinement: *DIF*4; data reduction: *REDU*4 (Stoe & Cie, 1992); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL*97.

Financial support from the Fonds der Chemischen Industrie is gratefully acknowledged. We thank Mr A. Weinkauf for technical assistance.



Figure 1

The molecule of the title compound in the crystal. Displacement ellipsoids are drawn at the 30% probability level. H-atom radii are arbitrary.



Figure 2

Packing diagram of the title compound, viewed parallel to the a axis. Secondary interactions are indicated by dashed bonds.

References

Flack, H. D. (1983). Acta Cryst. A39, 876-881.

- Hoechst (1966). Belg. Patent No. 667036, Farbwerke Hoechst AG; *Chem. Abstr.* (1966), **65**, 13741d.
- Jones, P. G., Kunze, C., Freytag, M. & Schmutzler, R. (2002). Acta Cryst. E58, 01434–01435.
- Kunze, C. (2002). PhD Thesis, Technical University of Braunschweig, Germany. [ISBN 3-89720-612-9]
- Kunze, C., Selent, D., Neda, I., Freytag, M., Jones, P. G., Schmutzler, R., Baumann, W. & Börner, A. (2002). Z. Anorg. Allg. Chem. 628, 779–787.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1994). XP. Version 5.03. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Stoe & Cie (1992). DIF4 and REDU4. Stoe & Cie, Darmstadt, Germany.